

**Course Description**

Prerequisite: CHEM 1002, 1212 or equivalent.

Principles of soil science; properties of soils related to plant growth and the environment.

Biology, chemistry and physics of soils. Includes genesis and classification, as well as widely practical applications of the basic sciences to crop production and environmental protection or restoration --soil survey, fertility, salinity, erosion and chemical pollution.

**Course Objectives**

Students will know and understand:

- 1 General components of soil and master soil horizons.
- 2 Basis for assigning color and its biological and/or chemical significance; definition of texture, how it is measured (mechanical analysis and feel method), and its importance in crop production and to the environment; types of structure, how management affects it and its importance; definitions of density and porosity, how to measure these parameters, effects of management on them and their importance.
- 3 That the chemical and physical weathering of rock yields particulate geologic materials that are altered in composition and disposition by the four processes of soil formation, which, in turn are controlled by the five factors of soil formation --parent material, climate, living organisms, relief, and time; major subordinate horizons.
- 4 Classification of soil by the USDA system, Soil Taxonomy, including hierarchy of the system, types of diagnostic horizons, and pedogenesis underlying the first hierarchal level, the 12 soil orders.
- 5 Types of information in hardcopy and web-based soil surveys and how to use this information.
- 6 Concepts of soil water capillarity; components of soil water potential energy; moisture characteristic curve; soil water movement based on gravitational and matric-pressure potential, including the effect of hydraulic conductivity; effects of soil horizonation on water flow; traditional physical and biological definitions of soil water based on the moisture characteristic curve; chemical and physical factors affecting the shape of the moisture characteristic curve, thus amount of plant-available water.

- 7 Processes in the field hydrologic cycle; water conservation and use management, including aspects of drainage and irrigation engineering.
- 8 Composition of soil air, importance of soil aeration, use of redox potential to measure aeration level, and management of soil aeration; definitions and environmental importance of wetlands and hydric soils.
- 9 Factors affecting soil temperature, and daily and annual cycles of temperature.
- 10 General types of soil colloids, the structure and properties of layer aluminosilicates, including types of charge, the phenomena and basis of ion exchange, soil-specific capacities for cation and anion exchange (CEC and AEC), and importance of these chemical reactions.
- 11 Definition of pH, its importance, sources of  $H^+$  and  $OH^-$ , pH buffering in soils, and how it is managed with base (e.g., lime) or acid (e.g., sulfur) amendments.
- 12 Effect of salinity on plant growth, definitions of saline, sodic and saline sodic soils, and reclamation strategies for soil with high salt or sodium content.
- 13 Relative numbers and biomasses of soil micro-flora and -fauna; rhizosphere effects on soil biology; types and importance of mycorrhizal fungi; important biochemical processes mediated by soil microfauna.
- 14 Carbon cycle and how alterations to it contribute to global climate change; decomposition rates of organic residues, including concepts of mineralization and immobilization, and factors affecting organic matter decomposition; composition of humus; importance of organic matter to crop production and environmental quality; composts and composting.
- 15 Nitrogen cycle, including mineralization and immobilization, ammonium fixation, ammonia volatilization, nitrification, nitrate leaching, denitrification, and biological N fixation; S cycle, including oxidation / reduction reactions, and environmental problems associated with S oxidation.
- 16 Phosphorus fertility limitations and the contrary problem of water quality degradation by P-induced eutrophication; fixation reactions of inorganic P, including factors affecting fixation; Importance of K to plants, sources of it and their relative plant-availability, excessive (luxury) uptake of K.
- 17 Micronutrient deficiencies and toxicities, including effects of pH and redox potential on availability of micronutrients; micronutrient fertilization, including salt and chelated forms, applied to soil or plant canopy.

- 18 Nutrient best management practices; types and uses of organic and inorganic fertilizers; Liebig concept of the limiting factor; nutrient management, including plant and soil analyses; fertilizer application methods; economic considerations.
- 19 Fate of organic and inorganic contaminants in the soil environment; sources of inorganic contaminants; remediation strategies, including bioremediation.
- 20 Accelerated water erosion, including environmental effects and economic costs; mechanics of water erosion, including types and use of USLE and related models in water erosion control; mechanics of wind erosion, including environmental and economic costs; mechanics of wind erosion, including types and use of WEQ and related models in wind erosion control.

### **Procedures to Evaluate Objectives**

1. Written exams over lecture and lab materials.
2. Quizzes over specific lecture and lab topics.

### **Detailed Topical Outline**

#### Lecture

#### Corresponding Labs

Introduction

Physical Properties

Mechanical analysis by pipette method; particle density

Bulk density, gravimetric water content, volumetric water content

Soil Formation

Classification

Profile descriptions at two sites of contrasting parent materials

Soil Survey

Use of soil survey book

Soil Water

Measure saturated hydraulic conductivity

Field Hydrologic Cycle

Aeration

Temperature

LectureCorresponding Labs

Colloidal Materials

Qualitative investigation of ion exchange and extraction of bases + acids for calculation of cation exchange capacity

pH Management

Qualitative investigation for presence of carbonates, assessing buffering, and quantitative determination of lime requirement

Salinity and Sodium

Biology

Enumeration of colony forming using serial dilutions

Organic Matter

Extraction of humic and fulvic acids, and qualitative investigation of their behavior with respect to ion exchange and coagulation

Nitrogen and Sulfur

Phosphorus and Potassium

Micronutrients

Nutrient Management

Mehlich 3 extraction for soil P, colorimetric analysis and use of results to make recommendations (includes N and K calculations)

Soil Pollution

Water and Wind Erosion

**Course Materials**

Text

The Nature and Properties of Soils, 11<sup>th</sup>, 12<sup>th</sup>, 13<sup>th</sup> or 14<sup>th</sup> Edition  
N.C. Brady and R.R. Weil

Lab Manual

Provided

Other Materials

<http://www.agronomy.lsu.edu/courses/agro2051/index.htm>

**Topics by Date, including Exams and Quizzes**

M	W	F
	<b>JANUARY 14</b> Introduction (Chapter 1)	<b>16</b> Introduction Physical Properties (Chapter 4)
<b>19</b> <b>MLK Day Holiday</b>	<b>21</b> Physical Properties	<b>23</b> Physical Properties Quiz 1 Introduction / Physical Properties
<b>26</b> Soil Formation (Chapter 2) <b>Lab 1 Texture</b>	<b>28</b> Soil Formation	<b>30</b> Soil Classification (Chapter 3) Quiz 2 Physical Properties / Soil Formation
<b>2 FEBRUARY</b> Soil Classification <b>Lab 2 Density</b>	<b>4</b> Soil Survey (Chapter 19)	<b>6</b> Soil Water (Chapter 5) Quiz 3 Soil Classification / Survey
<b>9</b> Soil Water <b>Lab 4 Soil Survey</b>	<b>11</b> Review Chapters 1, 2, 3, 4 and 19	<b>13</b> <b>EXAM 1</b>
<b>16</b> <b>Mardi Gras Holiday</b>	<b>18</b> <b>Mardi Gras Holiday</b>	<b>20</b> Field Hydrologic Cycle (Chapter 6) Quiz 4 Soil Water
<b>23</b> Field Hydrologic Cycle <b>Lab 5 Hydraulic Conductivity</b>	<b>25</b> Soil Air and Temperature (Chapter 7)	<b>27</b> Soil Colloids (Chapter 8) Quiz 5 Water Cycle / Air and Temperature
<b>2 MARCH</b> Soil Colloids <b>LAB EXAM 1</b>	<b>5</b> Review Chapters 5, 6 and 7	<b>7</b> <b>EXAM 2</b>
<b>9</b> Soil Reaction (Chapter 9) <b>Lab 6 Ion Exchange</b>	<b>11</b> Soil Reaction	<b>13</b> Salinity (Chapter 10) Quiz 6 Colloids / Reaction
<b>16</b> Soil Organisms (Chapter 11) <b>Lab 7 pH and Liming</b>	<b>18</b> Soil Organisms	<b>20</b> Organic Matter (Chapter 12) Quiz 7 Salinity / Organisms
<b>23</b> Organic Matter <b>Lab 8 Soil Biology</b>	<b>25</b> Review Chapters 8, 9, 10 and 11	<b>27</b> <b>EXAM 3</b>
<b>30</b> Macronutrients (Chapters 13 and 14) <b>Lab 9 Organic Matter</b>	<b>1 APRIL</b> Macronutrients Quiz 8 Organic Matter / Macronutrients	<b>3</b> <b>Spring Break</b>
<b>6</b> <b>Spring Break</b>	<b>8</b> <b>Spring Break</b>	<b>10</b> <b>Spring Break</b>
<b>13</b> Macronutrients <b>Lab Holiday</b>	<b>15</b> Micronutrients	<b>17</b> Nutrient Management (Chapter 16) Quiz 9 Macro- / Micronutrients
<b>20</b> Nutrient Management <b>Lab 10 Soil Fertility</b>	<b>22</b> Review Chapters 12, 13, 14 and 15	<b>24</b> <b>EXAM 4</b>
<b>27</b> Soil Pollution (Chapter 18) <b>LAB EXAM 2</b>	<b>29</b> Soil Erosion (Chapter 17) Quiz 10 Nutrient Management / Pollution	<b>1 MAY</b> <b>MAKE-UP EXAM 1, 2, 3 or 4</b>
<b>4</b>	<b>6</b>	<b>8</b> <b>EXAM 5</b> <b>7:30 – 9:30 AM</b>

**Location, Instructor Contact and Grading*****Where Held***

Lecture 134 Madison Sturgis  
 Lab 241 Madison Sturgis

***Contact***

Instructor Lewis Gaston  
 Office 224 Madison Sturgis  
 Lab 232 Madison Sturgis  
 Phone 578-1323  
 e-mail [lagaston@agctr.lsu.edu](mailto:lagaston@agctr.lsu.edu)  
 Hours 9:30 - 10:20 MW

***Grading***

<i>Scale</i>		<i>Components</i>	<i>Weight</i>
90	A	Lecture	
80	B		
70	C	Quizzes	10 %
60	D	Exam 1 – 4	40 %
Below	F	Exam 5	20 %
		Lab	
		Quizzes / Exercises / Participation	10 %
		Exam 1 – 2	20 %
		Total	100 %

***Comments on Exams, Quizzes and Make-Ups*****Lecture**

Lowest (one) mid-semester exam (1 – 4) dropped.

**ONLY ONE MAKE-UP EXAM AND IT IS ON THE LAST DAY OF CLASS.**

Lowest two quizzes dropped.

**NO MAKE-UP QUIZZES.**

**Lab**

**MAKE-UP EXAM TO BE ARRANGED WITH LAB INSTRUCTOR.**

**General Matters*****Code of Student Conduct***

It is the responsibility of all students to familiarize themselves with the Code of Student Conduct and other University rules and regulations governing student conduct and activities.

***Academic Dishonesty***

Academic dishonesty can result in probation, suspension, or expulsion from the course. For more information, refer to your handbook of responsibility in student university relationship or refer to the Code of Conduct that can be found in the Office of the Dean of Students website within the LSU home page.

***Disabilities Statement***

If you have special needs addressed by the Americans with Disabilities Act, please notify your instructor immediately for proper accommodations.